

Chemistry 103-6

Fall 2015

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Office Hours (in study room 1371C):	Mon 2:00-3:00PM; Wed 1:30-2:30PM
Lectures:	T/Th 2:30-3:45PM, Room 1351
Labs:	Room 2317
Course Website on Learn@UW:	https://learnuw.wisc.edu/
General Chemistry Homepage:	http://genchem.chem.wisc.edu/
Undergraduate Chemistry Office:	Room 1328 Chemistry 263-2424

INTRODUCTION

Introduction. Chemistry 103 is the first semester course in a two-semester General Chemistry sequence. The second semester course is Chemistry 104. Students who take Chemistry 103 often also plan to take Chemistry 104. Chemistry 103 and 104 provide a general background concerning the principles and factual basis of chemistry. The 103-104 sequence serves as a prerequisite for intermediate courses such as Organic Chemistry (341 or 343), Analytical Chemistry (327 or 329), and Inorganic Chemistry (311). Students in Chemistry 103 should have proficiency in college level algebra.

REQUIRED MATERIALS

1. *Chemistry: The Molecular Science* (5th Ed.), Moore and Stanitski. Although used copies may also be available, we have obtained a significantly discount price for the full bundle. You may purchase either the hardcover edition, a slightly less expensive unbound edition, or an electronic-only text (available with OWL2 account—see item 6 in this list).
2. *Chemistry 103 Laboratory Manual*, Fall 2015 and carbonless laboratory notebook. The manual and notebook can be purchased (cash only) outside the classroom during the first two week of classes and later at the first floor stockroom window (room 1334).
3. Safety goggles. Industrial quality eye protection is **required** in all chemistry laboratories. Safety goggles that fit over regular glasses can be purchased from University bookstores or along with the lab manual and notebook.
4. An electronic RF “clicker”. The lectures will make regular use of student “voting” on concept tests, surveys, and other questions. You will need to buy a radio-frequency clicker, specifically an I-clicker (not the web-clicker app) and bring it to every lecture. These can be purchased at the University Bookstore.

* Please sign any email messages to Dr. Block with your name, TA's name & your discussion or lab section number. For example: Angel Guzman, TA: Jaime Escalante, Sec. 442 (or 742).

5. An inexpensive calculator capable of calculating square roots, logarithms and exponential operations. The calculator will be used on exams, homework assignments, and in the lab. A programmable calculator may be used as long as no information is stored on it, such as chemical formulas or equations. It must be of the type allowable on an ACT or SAT exams (no cell phone or iPod calculators). You must clear the memory before entering the exam room.
6. An OWL2 account for access to on-line homework. This is bundled with your new textbook for no additional charge. Instructions for registering are given on the course homepage on Learn@UW. In brief, if you purchased a used textbook or received one from another student, **you must** purchase your own access to the OWL2 system after registering for the OWL2 course online. Register using the address: <http://login.cengagebrain.com/course/E-TWQN5ATJ4AUGD>. Use your wisc.edu e-mail address as your primary e-mail and your NetID in the “student ID” field.
7. USB Drive: A USB flash drive that will hold at least 2 GB is highly recommended for laboratory data collection.

COURSE INFORMATION

Course Organization and Expectations. This course is designed to help you to learn chemistry. Your lecturer and TA will do their best to guide you in mastering the material, but no course or instructor can learn for you. Learning is something only you can do. Many learning activities are offered in order to meet the needs of different types of students; however, if you find that your learning needs are not being met or you are not satisfied with some aspect of the course, please bring your concern to your lecturer or your TA.

Many of you are first semester freshmen. You will find several significant differences between your high school course and this course. Perhaps the biggest is the amount of time you should expect to put into this course, ranging from 8-12 hours of outside-of-class studying per week. The precise amount will depend on your academic background, native ability, and desired success level. In order to keep up, it is essential that you work on improving your studying and time management skills. A recommended study strategy for this course is: 1) read through textbook sections before each lecture, 2) attend class and take your own notes, 3) begin to work homework problems as soon as possible after reading the chapter. When you encounter problems that you cannot solve, refer to the text and its example problems, your notes, a tutorial, or your fellow students. Forming a study group to work through problems is an excellent way to learn chemistry. You will find summary problems, key terms, learning goals, and plenty of additional problems at the end of each chapter. These lists will help you focus on primary themes and gain practice with the material.

Throughout this course emphasis will be placed on understanding chemistry and learning to think effectively in solving problems. Successful problem solving requires a basic knowledge of principles, facts, and terms: a vocabulary of chemistry. This course includes a range of activities that are aimed at facilitating the learning process. These activities are described below.

Lectures. During lectures I will introduce principles and illustrate concepts with examples and demonstrations. In addition to your notes, a set of lecture notes taken in class by a TA will be available at our Learn@UW web site listed above about a day after the lecture. Each lecture will also contain “clicker” questions. Your participation in these questions will count toward your final grade. In order to earn full points for clicker participation, you must answer at least 80% of in-class clicker questions.

Classroom etiquette does become important with courses this large. Cell phones should be turned off or at least silenced. While laptops are not prohibited in class, you will not have any need for them during lecture. Using the computer or other devices during class for activities not related to the class is very distracting, not only for you but for those who are sitting nearby. Finally, our lecture room desks are very noisy when raised or lowered; so please wait until the instructor is completely done speaking before you lower your desk at the end of class. As much as possible class will be dismissed at 3:45, but sometimes just another minute or two is needed to finish up. Please be considerate of your classmates.

Discussion Section. Twice a week, you will meet with a TA and your classmates for discussion. In these meetings, you will discuss assigned homework problems, work with groups of students to learn new material or reinforce/review existing ideas, learn about upcoming laboratory assignments, and have a forum for answering questions. *Please* prepare for discussion by bringing specific questions to class – this is a great opportunity to learn from your TA and fellow classmates. In addition, there will be a few ungraded problems assigned from the textbook each week. You will have the opportunity to pick one of those for your TA to walk through with you. Lastly, each discussion section will have the opportunity to assemble feasible exam problems, which will then be posted for the class to see. A number of these questions will be chosen for inclusion on each exam.

Problem Sets. Problem solving is a crucial aspect of this course and problems will be assigned on a regular basis. These will be completed online via the OWL2 homework system. **A subset of the problems will be required, occasionally supplemented with additional recommended (but optional) practice problems.** In addition, each problem set will have a 10 additional points available. For example, the maximum score for a homework set may still be 100 points, but 110 points will be available on the assignment. The additional points can offset small errors and difficulties associated with the OWL2 system user learning curve. The system gives hints and allows multiple attempts, each with feedback. A small deduction (detailed for each problem set in the assignment and problem descriptions) is taken for each successive attempt. You can log on multiple times to complete the assignment. See Learn@UW for more information on the OWL2 online homework system. Assignments will typically be due by 11:55PM each Sunday night.

If you encounter technical difficulties with OWL2 pertaining to how answers are submitted/accepted or why you did not get credit for an answer that was later revealed to be correct, please send an e-mail to chem103hw@chem.wisc.edu with your name, course number (103), lecture section (6), and a brief description of your difficulty. The group of people who assist you will not answer content related inquiries.

Your textbook is an excellent source of additional practice problems, and answers to selected problems are given at the back of the book. Bring questions to your discussion section and to TA and faculty office hours. *In order to excel in this course you **must** solve problems. Lots of them.*

Quizzes. Approximately 10-12 fifteen-minute quizzes will be posted online at the start of each week to help you evaluate your progress. These quizzes will not count toward your final grade, except that their completion is part of your discussion participation grade. Your TA will go over the quiz in discussion, and **you** will grade your own quiz. Though they are not graded, you should use your score as an indication of your progress in the course.

Lecture Demonstrations. We will use demonstrations during lecture to illustrate important ideas and facts. Be sure to make careful observations of what happens. Questions about observations or principles that have been presented via demonstrations may appear on exams.

Exams. There will be three in-class exams of 75 minutes each and one two-hour final exam. **No** makeup exams will be given. Exams may include questions based on the laboratory material. The final exam will cover material from the entire semester. **Please be alert to these exam dates.** You must report any religious conflicts with exams or laboratory exercises to your teaching assistant within the first two weeks of classes.

Exam Dates: Tuesday, September 29	2:30 – 3:45PM
Tuesday, October 27	2:30 – 3:45PM
Tuesday, November 24	2:30 – 3:45PM
Final Exam: Friday, December 18	5:05PM – 7:05PM

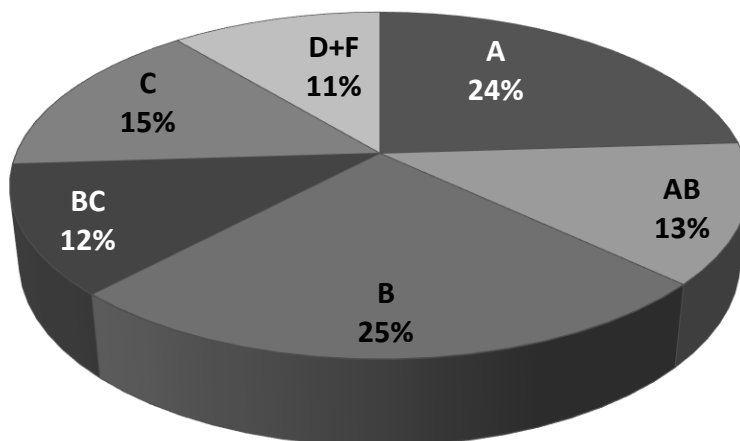
Grades. Your final grade will be computed with the following scheme:

Three 75-minute exams	12% each
Online Homework	15%
Laboratory	20%
Discussion participation	4%
Clicker participation	3%
<u>Final Exam</u>	<u>22%</u>
TOTAL	100%

Your scores are always available to you at Learn@UW. There are no opportunities for extra credit beyond the couple points on assignments.

The approximate distribution of final grades is given below. It is important to note that the distribution will be adjusted upwards if class performance exceeds expectations. For example, we guarantee that at least 24% of the grades will be A, and it may be slightly higher.

Approximate Distribution of Final Grades



LABORATORY

The laboratory experiments are a vital part of this course; you will develop skills that are not easily learned or demonstrated in lectures. These skills include:

- Designing experiments and interpreting data
- Using laboratory equipment properly
- Working with your fellow students in the laboratory
- Communicating your ideas about the data through discussions and writing

You must successfully complete all of the laboratory assignments to receive a passing grade in this course.

Lab Preparation. You **must** prepare in advance for each laboratory exercise by writing an introduction and procedural outline in your lab notebook. During the lab period you will carry out the experiment, take notes, and complete your data analysis. All your work **must** be turned in at the end of the period in the form of the duplicate pages from your lab notebook. You will be graded on your pre-lab preparation, in-lab experimental technique and data analysis, and on your note taking skills. Your laboratory report is almost always due at the end of the laboratory period. Late laboratory reports are not graded. The lab schedule is printed on the attached calendar. Exercises in italics are computer labs.

Please note that sandals are not acceptable footwear in the laboratory. Contact lenses should **not** be worn in the laboratory because fumes or splashes may be caught between them and your eye. Further attire requirements are described in your laboratory manual and by your TA.

Attendance. You must attend all laboratory sessions. There is no opportunity to make up a laboratory that you miss; a grade of zero will be recorded for unexcused absences. If you have an excuse for missing lab, notify your TA as soon as possible, preferably before the lab period.

Health or Disability Concerns. If you have special needs, please make an appointment to speak to your lecturer and TA at your earliest convenience.

ADDITIONAL RESOURCES

Numerous resources are available to assist you with either this course in particular or college life in general. It is up to you to take advantage of these resources to ensure your success both in this course and at UW-Madison.

Course Web-site on Learn@UW (<https://learnuw.wisc.edu/>): Our course website can be accessed via Learn@UW. The syllabus, schedules, office hours, TA lecture notes, course handouts, announcements and grades will all be available on Learn@UW.

General Chemistry Web Site (<http://www.chem.wisc.edu/content/genchem-main/>): Resource materials for general chemistry students are available on the General Chemistry website. The computer laboratory exercises, ChemPages, and other lab resources are accessed via the "Materials for Labs" link.

Study Groups: You may collaborate with other students on homework assignments and laboratory discussion questions. Study groups reflect the teamwork inherent in the way modern science is done; scientists frequently collaborate with others, either within the same department or at a distance with persons in other cities, states or countries. It is important to realize that although you may collaborate with other students on assignments, the work you submit must be your own.

Tutoring Services: A number of tutoring resources are available on campus, some free and some for a fee. For more information, see our Learn@UW site or the General Chemistry home page (<http://www.chem.wisc.edu/content/genchem-main/>) under the "Information for Students" section.

Students with Disabilities: Appropriate accommodations for lecture, laboratory, discussion, and/or exams can be arranged for students with disabilities. The McBurney Disability Resource Center (<http://www.mcburney.wisc.edu/>) can provide assistance. Accommodations still must be made well in advance, so please pursue these avenues immediately.

Advising and Counseling Services (University Health Services): College life can be stressful. If you are struggling with your academic course load or other academic issues, your advisor is a good resource. If you are struggling emotionally with anxiety, depression, or other health issues, individual counseling is available at University Counseling and Consultation Services. For more information go their website (<http://www.uhs.wisc.edu/>) or call 265-5600. Crisis intervention services are also available 24 hours a day by dialing this same phone number and pressing option 9.

Academic Misconduct: It is expected that all students will conduct themselves with honesty, integrity, and professionalism. Any student caught cheating on an exam (including submitting an altered exam for regrade) will receive an F in the course. Any student caught cheating on homework, a quiz, or lab (for instance, copying another person's work or fabricating data) will receive a zero for that assignment. A second infraction will result in an F for the course. More information on what constitutes academic misconduct and policies on handling misconduct can be found in your chemistry lab manual and at the following website:

<http://www.wisc.edu/students/saja/misconduct/UWS14.html>

Course Outline and Calendar

The course outline appears on the next page. Dates for lecture topics are **approximate**. The exam dates are **fixed**. The course website on Learn@UW will have all specific reading suggestions and due dates as they become available.

Week	Date	Topic	Chapter	Laboratory Exercise
1	R 9/3	Matter and Measurement	1	
2	T 9/8	Atoms and Elements	2	Citizenship in the Lab
	R 9/10	Atoms, Molecules, Ions	2	
3	T 9/15	Stoichiometry	2	Solutions, Density, and Graphing
	R 9/17	Chemical Reactions	3	
4	T 9/22	Chemical Reactions and Stoichiometry	3	<i>Reaction Types and Chemical Logic</i>
	R 9/24	Stoichiometry	3	
5	T 9/29	In Class Exam		No Lab
	R 10/1	Energy & Enthalpy	4	
6	T 10/6	Energy & Enthalpy	4	Zinc and Iodine
	R 10/8	Energy & Enthalpy	4	
7	T 10/13	Atomic Structure	5	Solution Calorimetry
	R 10/5	Atomic Structure & Multi-electron Atoms	5	
8	T 10/20	Periodic Properties	5	Synthesis of an Alum
	R 10/22	Periodic Properties & Review	5	
9	T 10/27	In Class Exam (Drop Deadline Oct 30)		No Lab
	R 10/29	Chemical Bonding	6	
10	T 11/3	Chemical Bonding	6	Light, Color, and Solutions
	R 11/5	Molecular Orbital Theory	6	
11	T 11/10	Molecular Geometry	7	<i>Molecular Geometry and WebMO</i>
	R 11/12	Molecular Geometry	7	
12	T 11/17	Gases	8	Project Lab
	R 11/19	Gases	8	
13	T 11/24	In Class Exam		No Lab
		Thanksgiving Break		
14	T 12/1	Solids	9	<i>Window on the Solid State</i>
	R 12/3	Intermolecular Forces	9	
15	T 12/8	Intermolecular Forces & Phase Diagrams	9	No Lab
	R 12/10	Liquids, Solids, and Phase Diagrams	9	
16	T 12/15	Wrap-up and review		
	F 12/18	Final Exam 5:05-7:05PM	All	