

CHEMISTRY 329: Fundamentals of Analytical Science

Spring 2015

Lecturer: Dr. Linda Zelewski
E-mail: zelewski@wisc.edu (Please sign your email messages with your name, your TA's name and your lab section.)
Office: Chemistry 7108 (Take the elevator in the lobby at the corner of Johnson St. and Charter St. up to the 7th floor.)
Office Hours: See Learn@UW
Course Credit: 4 credit hours
Lecture 1: MW 8:50-9:40 a.m. in B371 Chemistry
Discussion: F 8:50-9:40 a.m.
Lab: MW 1:20-5:25 p.m.
Web Site: <https://learnuw.wisc.edu/>

Laboratory Director: Dr. Pam Doolittle (chem 2303A, pam@chem.wisc.edu)

REQUIRED MATERIALS

Textbook: *Quantitative Chemical Analysis*, Eighth Edition, by Daniel C. Harris, W.H. Freeman and Company.

Lab Manual: *A Manual of Experiments for Analytical Chemistry, Spring 2015*, Department of Chemistry, UW-Madison. Lab manuals will be sold by Alpha Chi Sigma in the lobby of the chemistry building (Mills Street and University Avenue) beginning January 20.

Lab Notebook: Carbonless laboratory notebook with duplicate, numbered pages.

Calculator: A scientific or graphing calculator is required. Only calculators that are permitted on SAT or ACT tests may be used on exams. You may NOT use any stored information, programs, or applications on exams unless given explicit permission.

Safety Goggles: Industrial quality eye protection is required at all times when you are in the lab. Safety goggles that completely seal around the eyes and fit over regular glasses can be purchased from local bookstores.

USB Drive: A USB flash drive that will hold at least 2 GB is required for laboratory.

Course Web Site: Assignments, announcements, lecture notes, handouts and homework will be posted on our course web site. You will also use the web site to complete online Pre-lab Quizzes and to view your grades.

SUGGESTED MATERIALS

Solutions Manual for Quantitative Chemical Analysis, Eighth Edition, by Daniel C. Harris, W.H. Freeman and Company. (Also available in the Chemistry Library.)

COURSE ORGANIZATION

LECTURE

Lectures organize the material, cover the basic principles of each topic, and present illustrations and demonstrations. A lecture is not intended to describe or explain everything you should learn; rather, it will indicate what topics it is important to study and provide some insight into those topics. Read the assigned sections of the textbook prior to lecture. Take notes during lecture to capture your understanding of what you heard and saw. After lecture, reread and study the appropriate pages in the textbook.

Lecture Notes

A set of lecture notes, taken by one of the TAs, is available on Learn@UW. These notes are relatively complete; however, they will not replace your notes or the textbook.

DISCUSSION

Attendance is required. Your TA will provide important pre-lab information as well as answer any questions you may have about labs, homework, or lecture material. To get the most from discussion section, you need to be prepared. It is your responsibility to communicate to your TA the concepts you do not understand and the skills you need to practice. You should ask your TA specific questions and make sure you understand the questions and the answers given by your TA and by fellow students. Bring your lab notebook and manual with you to discussion.

CLASSROOM ETIQUETTE

Cell phones should be turned off. While laptops are not prohibited in class, you will not have any need for them during lecture. Using the computer or other devices during class for activities not related to class (such as surfing the web, playing video games, texting, etc.) is both rude and distracting, not only for you, but for those who are sitting nearby.

HOMEWORK

There will be nine homework assignments. The tentative date each homework assignment will be given is shown on the lecture schedule. Each assignment will have two parts. The first part is a list of problems from the textbook. These problems will not be collected or graded since worked out solutions to these problems can be found in the *Solutions Manual for Quantitative Chemical Analysis*. The second part of the assignment will have problems that are to be worked out and turned in for grading.

When doing homework assignments, you may work independently or collaborate with others. If you choose to collaborate, you must still work out and hand in your own solutions. If you worked with other students, you must indicate on your paper who you worked with. In order to receive credit for homework you must clearly communicate your thought process and show all of your calculations. Write your TAs name and your section number on your assignment.

Homework is due at the BEGINNING of the specified period. To discourage late submissions, 4 points will be deducted from homework turned in later the same day and before 5:00 p.m. Homework turned in later than this will receive a zero. **Late homework assignments must be handed directly to your TA or placed in Dr. Zelewski's mailbox before 5:00 p.m.** If there are extenuating circumstances, such as a serious illness or family emergency, please email Dr. Zelewski (and copy your TA) as soon as possible. Dr. Zelewski may consider an extension for extenuating circumstances.

LABORATORY

Analytical chemistry is an experimental science and the laboratory is a significant part of the course. You will perform twelve standard labs involving chemical analyses aimed at teaching you specific skills. The procedures for these experiments are provided in *A Manual of Experiments for Analytical Chemistry*. You will also spend approximately 5 weeks designing and conducting your own experiments for a Laboratory Project.

YOU MUST COMPLETE ALL OF THE LABS AND EARN A MINIMUM SCORE OF 60% ON ALL LABORATORY ASSIGNMENTS (COMBINED PRELAB QUIZ AND LAB REPORT GRADES) IN ORDER TO RECEIVE A PASSING GRADE IN THE COURSE.

Laboratory Project

For the Laboratory Project, you will be analyzing phosphorus concentrations in lake water and investigating factors that influence its concentration. You will work in a small group of four or five students and have ten laboratory periods to conduct your experiments. Your group will present your results in both a written report and an oral presentation before the course instructors. More information regarding the project will be provided during the semester.

Standard Labs

Each of the standard labs is worth 15 points and has three graded components. You must take the online Pre-lab Quiz on Learn@UW prior to coming to lab (6 points). You will be graded on the accuracy of your lab results (5 points), and your lab notebook will be graded for completeness and clarity (4 points). ***Please note that for some weeks each lab section has a different lab schedule.*** Be sure to check the schedule at the end of this syllabus so that you prepare for the appropriate lab and take the appropriate online Prelab Quiz.

Pre-lab Quiz (for Standard Labs)

Prior to taking a quiz, read about the experiment in the lab manual and prepare your notebook pages. You should be familiar with the overall concepts of the experiment and the methods used in the experiment. You should also understand the procedure and how to do the calculations you will need to do with the raw data.

You may attempt the quiz twice and your highest score will appear in the grade book. Note that there is a time limit (usually 30 minutes) for each attempt. The clock begins timing once you start the quiz and you cannot stop and come back later in the day to finish, so be prepared to complete the entire quiz once you begin. Most quizzes have questions that involve calculations, so you should have a calculator, scratch paper, pencil, and your lab manual available when you attempt a quiz.

Check your lab schedule to make sure you are attempting the correct quiz. Quiz ending dates and times have been set-up by section and are set to end at 1:15 p.m. the day you are scheduled to do a lab. ***Quizzes must be completed before this time or you will receive zero points for the prelab quiz.*** Attempt taking the prelab quiz at least 24 hours before the scheduled deadline. UW-Madison network maintenance is periodically scheduled during the hours of 10:00 p.m. to 8:00 a.m., during which time Learn@UW is not available. Scheduled maintenance is not an excuse for not completing the prelab quiz on time.

When taking a quiz, ***do not include units when entering numbers for computational questions.*** The auto-grade feature will not recognize the units and it will mark your answer wrong. In all

other parts of the course (lab notebook, homework, exams, etc.) you should always include units. Also be sure to enter the correct number of significant figures for a calculated answer. ***An error in the number of significant figures will result in an incorrect answer and zero points for your answer.***

Please notify the laboratory director, Dr. Doolittle (pam@chem.wisc.edu), or Dr. Zelewski (zelewski@wisc.edu) promptly if you can't access your quiz and/or if your ending date or time is incorrect for a particular quiz. Make sure to include your TAs name and laboratory section number in your email message.

Laboratory Notebook

Your lab notebook should include the following:

- Overview of the experiment or purpose statement
- Experimental procedure
- Relevant chemical reactions
- Raw data from all measurements
- One complete sample calculation including units and proper significant figures
- Results and summary section

The first three items and tables for the raw data should be prepared ahead of time. Sample calculations should be outlined ahead of time and doing so may help you with the prelab quiz. Your results and summary section should be brief. In addition to your final results, it should include comments on whether your data are reasonable and/or any problems that occurred that could affect your results. Someone else should be able to repeat the experiment based on what you've written in the notebook. For more guidelines on proper record keeping, consult the lab manual and section 2-2 of the textbook.

Laboratory Reports

Your laboratory report will consist of the carbonless copies of the relevant pages from your laboratory notebook and the completed (yellow) summary sheet from your lab manual. You will be graded on the accuracy and precision of your results and the quality of your notebook record. See "Laboratory Report Grades" on Learn@UW for more detailed information on how laboratory reports will be graded.

Laboratory reports are due to your TA at the beginning of the following laboratory period after you have completed the experiment. A penalty of one point per day will be deducted if you submit your report late. Late lab reports placed in your TA's mailbox must be accompanied by an email message sent prior to submitting the report. Failure to email your TA may result in additional points lost if your TA does not know to look for your report in his/her mailbox.

Laboratory Conduct

Safety goggles and proper attire must be worn at all times in the laboratory. Your lab begins at 1:20 p.m., and it is essential that you arrive on time so that you don't miss important information. Points will be deducted from your lab score for unsafe or inadequate conduct including, arriving late, not wearing goggles, not cleaning up spills promptly, and not cleaning your lab area before leaving the lab.

GRADES

Intended Grading Scale

Letter grades will be assigned at the end of the semester based on the following intended grading scale:

A	90.0%
AB	88.0%
B	80.0%
BC	78.0%
C	70.0%
D	60.0%

This scale *may* be adjusted downward at the end of the semester, depending on the overall class average. It will never be adjusted upward.

You will earn points based on how well you do on exams, homework and labs. Below are the tentative point values of items that will be graded. Adjustments to graded items and point values may be made during the semester if needed.

3 Midterm Exams @ 150 points each	450 points
9 Homework Assignments	~225 points
Instructional Prelab Quiz	6 points
12 Laboratory Reports & Prelab Quizzes	180 points
Laboratory Project	90 points
Discussion Participation	15 points

If no changes are made to the point values above, the total possible points at the end of the semester will be 966 points.

Your letter grade will be determined by calculating your final percentage using the formula:
 $\% \text{ score} = (\text{total points earned} / \text{total possible points}) \times 100\%$.

Exams

Three two-hour midterm exams will be given. The first two exams will be given during the laboratory period and the third exam will be given during the final exam period.

Exam 1:	Wednesday, February 25 during the lab period (room TBA)
Exam 2:	Wednesday, April 8 during the lab period (room TBA)
Exam 3:	Thursday, May 14, 12:25 to 2:25 p.m. (room TBA)

Exams are cumulative in the sense that many of the concepts covered in this course build on one another and a good understanding of earlier material is required for mastering later material. Objectives for each exam will be posted on Learn@UW one week prior to the exam.

Review Your Grades

Your grades will be available on Learn@UW. Be sure to review your scores regularly and notify your TA promptly of any discrepancies. Do not wait until the end of the semester to request corrections.

ACADEMIC MISCONDUCT

It is expected that all students will conduct themselves with honesty, integrity, and professionalism. Any student caught cheating on an exam will receive an F in the course. This penalty includes incidents such as looking at another student's paper during an exam or altering an exam after it has been graded and then submitting it for re-grading. Any student caught cheating on a lab or homework assignment (for instance, copying another person's work or fabricating data) will receive a zero for that assignment. A second infraction will result in an F for the course. More information on what constitutes academic misconduct and UW policies on handling misconduct can be found at:

<http://www.wisc.edu/students/saja/misconduct/UWS14.html>.

ATTENDANCE POLICY

Your attendance at all scheduled classes (lecture, discussion and lab) is mandatory and essential for success in the course. However, circumstances occasionally occur where you may need to miss a class.

Planned Absences

If you need to miss class for a religious observance, a UW athletic commitment, graduate school interview or some other legitimate reason, you must make arrangements to make up the missed work *a minimum of one week before the absence occurs*, otherwise, the absence will be unexcused and you will receive a zero on the missed work. *Homework must be turned in early (before the deadline) if you will be absent on the day it is due.*

Serious Illness/ Family Emergency

If you are seriously ill or experiencing a family emergency and are unable to attend lab, inform your TA via email and copy Dr. Zelewski. If you need an extension on your homework due to a serious illness or family emergency, arrangements must be made with Dr. Zelewski BEFORE the homework due date. Late homework cannot be accepted for any reason once the answer key is posted.

STUDENTS WITH DISABILITIES

Students with documented disabilities (McBurney students) or any special concerns should contact Dr. Zelewski as soon as possible so that appropriate accommodations can be arranged.

SUGGESTIONS FOR SUCCESS

Many students find Chemistry 329 to be both challenging and rewarding. It is a four-credit course and you can expect to work hard. As an experienced college student, you have likely developed a style for studying and learning that works well for you. Below are some additional tips that if followed help many students succeed in this course.

- Attend all lectures, labs and discussions.
- Read the related material in the textbook *before* lecture. Some students find it helpful to take notes on what they've read.
- The textbook is not a novel! If you try to read it as such, you will likely fall asleep. The trick to successfully reading a technical book is to be an "active" reader. Have paper and pencil nearby and use it to write out solutions or solve equations. Try working examples without looking at the solutions.
- Review your notes after lectures. Reread the related material in the textbook. If there are parts you don't understand, seek help from an instructor or classmate.
- Solve lots of problems! Do all of the homework, plus extra practice problems. You will become more proficient and do better on exams if you have worked through lots of problems.
- If you often work in a group to do homework problems, be sure to balance that time with independent problem solving. You won't have the group with you during exams!
- Make good use of instructors' office hours.
- Seek help promptly if you are confused or have questions. Your confusion will only be compounded by letting it slide. Keep up with the material as last minute cramming is not effective.

CHEMISTRY 329-1 LECTURE OVERVIEW*

WEEK	DAY & DATE	LECTURE TOPIC	CHAPTERS	HW
1	W Jan 21	The Analytical Process	0, 2	
2	M Jan 26 W Jan 28	Chemical Measurements Significant Figures, Uncertainty	1 3, Appendix C	HW#1
3	M Feb 2 W Feb 4	Statistics Statistics	4 4	HW #2
4	M Feb 9 W Feb 11	Spectrophotometry Laboratory Project: Analysis of Phosphorus Concentrations in Lake Mendota	17	
5	M Feb 16 W Feb 18	Quality Assurance & Calibration Methods Spectrophotometry	5 17, 19	HW #3
6	M Feb 23 W Feb 25	Spectrophotometry Chemical Equilibrium EXAM 1 (during the lab period)	19, 20 6	HW #4
7	M Mar 2 W Mar 4	Ionic Strength and Activity Systematic Treatment of Equilibrium	7 7	
8	M Mar 9 W Mar 11	Monoprotic Acid-Base Equilibria Monoprotic Acid-Base Equilibria	8 8	HW #5
9	M Mar 16 W Mar 18	Polyprotic Acid-Base Equilibria Polyprotic Acid-Base Equilibria	9 9	
10	M Mar 23 W Mar 25	Acid-Base Titrations Acid-Base Titrations	10 10	HW #6
11	Mar 30-Apr 3	SPRING BREAK		
12	M Apr 6 W Apr 8	EDTA & Other Complexes EDTA & Other Complexes EXAM 2 (during the lab period)	11 11	HW #7
13	M Apr 13 W Apr 15	Electrochemistry Electrochemistry	13 13	HW #8
14	M Apr 20 W Apr 22	Electrochemistry Electrochemistry	14 14	
15	M Apr 27 W Apr 29	Separations Gas Chromatography	22 23	HW #9
16	M May 4 W May 6	High-Performance Liquid Chromatography Finish up and review	24	
FINALS	R May 14	EXAM 3 (Thursday, May 14 from 12:25-2:25 p.m.)		

*See Learn@UW for specific reading, homework, and laboratory assignments and due dates.

Week	Date	Sec 621	Sec 622	Sec 623	Sec 624
1	19-Jan	<i>No Lab -- Martin Luther King Jr. Day</i>			
	21-Jan	Check-in/Weighing	Check-in/Weighing	Check-in/Weighing	Check-in/Weighing
2	26-Jan	Volumetric Calibration	Volumetric Calibration	Volumetric Calibration	Volumetric Calibration
	28-Jan	Standardization of HCl	Standardization of HCl	Standardization of HCl	Standardization of HCl
3	2-Feb	Standardization of NaOH	Standardization of NaOH	Standardization of NaOH	Standardization of NaOH
	4-Feb	Determination of % KHP in a Mixture	Determination of % KHP in a Mixture	Determination of % KHP in a Mixture	Determination of % KHP in a Mixture
4	9-Feb	Spectrophotometric Det. of Fe	Spectrophotometric Det. of Fe	Spectrophotometric Det. of Fe	Spectrophotometric Det. of Fe
	11-Feb	Project Introduction	Project Introduction	Project Introduction	Project Introduction
5	16-Feb	Project	Project	Project	Project
	18-Feb	Project	Project	Project	Project
6	23-Feb	Project	Project	Project	Project
	25-Feb	EXAM 1			
7	2-Mar	Chemical Oxygen Demand	Chemical Oxygen Demand	Chemical Oxygen Demand	Chemical Oxygen Demand
	4-Mar	A Study of Bromocresol Green	A Study of Bromocresol Green	A Study of Bromocresol Green	High Pressure Liquid Chromatography
8	9-Mar	Adventures with Buffers	Adventures with Buffers	High Pressure Liquid Chromatography	A Study of Bromocresol Green
	11-Mar	ID of Unknown Acid	High Pressure Liquid Chromatography	Adventures with Buffers	Adventures with Buffers
9	16-Mar	High Pressure Liquid Chromatography	ID of Unknown Acid	ID of Unknown Acid	ID of Unknown Acid
	18-Mar	Project	Project	Project	Project
10	23-Mar	Project	Project	Project	Project
	25-Mar	Project	Project	Project	Project
11	30-Mar	SPRING BREAK			
	1-Apr	SPRING BREAK			
12	6-Apr	Project	Project	Project	Project
	8-Apr	EXAM 2			
13	13-Apr	Project	Project	Project	Project
	15-Apr	Project	Project	Project	Project
14	20-Apr	Silver Electrode Study of Equilibria	Fluoride Ion Electrode	Silver Electrode Study of Equilibria	Gas Chromatography
	22-Apr	Fluoride Ion Electrode	Silver Electrode Study of Equilibria	Gas Chromatography	Silver Electrode Study of Equilibria
15	27-Apr	Spectrophotometric Study of Fluorescein	Gas Chromatography	Fluoride Ion Electrode	Spectrophotometric Study of Fluorescein
	29-Apr	Gas Chromatography	Spectrophotometric Study of Fluorescein	Spectrophotometric Study of Fluorescein	Fluoride Ion Electrode
16	4-May	<i>Project Presentation</i>	Check Out	<i>Project Presentation</i>	Check Out
	6-May	Check Out	<i>Project Presentation</i>	Check Out	<i>Project Presentation</i>