



### Class Synopsis

Fundamental ideas of thermodynamics and statistical mechanics will be discussed. These include a brief review of thermodynamics, introduction of various ensembles, methods for computing thermodynamic observables of molecular systems using microscopic models, basic theories of phase transitions, imperfect gases, liquids and polyelectrolytes. If time permits, topics related to interface and polymer dynamics will also be covered.

Homework problems are largely of the pencil & paper type. There will be two mid-term exams and one cumulative final exam.

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**Lectures:** Chem 8335 MWF 11:00AM-11:50AM

### Textbook and [References](#)

- *Introduction to Modern Statistical Mechanics*, David Chandler, Oxford University Press, 1987 [An elegant basic introduction]
- *Statistical Mechanics*, Donald A. McQuarrie, University Science Books, 2000 [A classic text]
- *An Introduction to Statistical Thermodynamics*, T. L. Hill, Dover, 1986 [A classic text that includes clear explanations on key topics]
- *Molecular Driving Forces: Statistical Thermodynamics in Chemistry and Biology*, Ken Dill and Sarina Bromberg, Garland Science, 2010 [A wonderfully clear introduction with many chemical and biological applications]
- *Basic concepts for simple and complex liquids*, Jean-Louis Barrat and Jean-Pierre Hansen, Cambridge University Press, 2003 [A collection of somewhat advanced topics related to simple and complex liquids]

	<b>Lec</b>	<b>Discussion Topic</b>	<b>Reference</b>
Sept. 4	L1	<b>Goal; Rev. Thermo (1st Law)</b>	Chandler, Ch1
Sept. 6	L2	<b>2nd Law and equilibrium condition</b>	Chandler, Ch1
Sept. 9	L3	<b>Legendre transform; Maxwell relation</b>	Chandler, Ch1
Sept. 11	L4	<b>Euler Theorem; Extensive variables; Gibbs-Duhrem</b>	Chandler, Ch1; Dill, Ch 9
Sept. 13	L5	<b>Equilibrium Statistical Thermodynamics; probabilities I</b>	Chandler, Ch1
Sept. 16	L6	<b>Probabilities II</b>	Dill, Ch1; Kardar Ch 2
Sept. 18	L7	<b>Probabilities III</b>	Dill, Ch1; Kardar Ch 2
Sept. 20	L8	<b>Ensembles I: microcanonical</b>	Chandler, Ch3
Sept. 23	L9	<b>Ensembles II: canonical I</b>	Chandler, Ch3
Sept. 25	L10	<b>Ensembles III: canonical II</b>	Chandler, Ch3
Sept. 27	L11	<b>Ensembles IV: Generalized ensembles</b>	Chandler, Ch3
Sept. 30	L12	<b>Grand-canonical ensemble; Ideal gas law from uncorrelated fluctuati</b>	Chandler, Ch3
Oct. 2		<b>Ideal System I: Photon gas, Bose-Einstein/Fermi-Dirac distributions</b>	
Oct. 4	L13	<b>Ideal System II: Classical limit, monoatomic gas; Gibbs paradox</b>	Chandler, Ch4
Oct. 7	L15	<b>Diatomics II; Equipartition; Classical expression</b>	Pathria
Oct. 9	L16	-----	McQuarrie, Ch6,7
Oct. 11	L17	<b>Chemical equilibrium</b>	Chandler, Ch4
Oct. 14	L18	<b>Lattice vibration I - NMA</b>	
Oct. 14 ev	L19	<b>Lattice vibration II - Einstein vs. Debye</b>	McQuarrie, Ch11
Oct. 18		<b>Mid-term Exam 1 (thermo; probabilities and ensembles)</b>	
Oct. 21		<b>Review of the first mid-term</b>	
Oct. 23	L20	<b>Phase Transition I - the Ising model, transfer matrix</b>	Chandler, Ch5; Yeomans, Ch1
Oct. 25	L21	<b>Phase Transition II - 1D vs. 2D and <math>T_c</math> [<math>\langle M \rangle</math>]</b>	Chandler, Ch5
Oct. 28	L22	<b>Phase Transition III - Symmetry breaking, Correlation &amp; susceptibili</b>	Chandler, Ch5
Oct. 30	L23	<b>Phase Transition IV - Mean Field theory</b>	Chandler, Ch5
Nov. 1	L24	<b>Phase Transition V: Variational &amp; perturbation theory</b>	Chandler, Ch5
Nov. 4	L25	<b>Co-operativity: Helix-coil transition</b>	Dill; Ch26
Nov. 6	L26		
Nov. 8		-----	
Nov. 11	L27	<b>Gibbs phase rule; Clausius-Clapeyron; interface; Gibbs adsorption isc</b>	Chandler, Ch2
Nov. 13	L28	<b>Lattice model for liquids and solution mixtures</b>	Chandler, Ch2
Nov. 15	L29	<b>Lattice model for transfer between phases; another look at chemical</b>	Dill, Ch14-15
Nov. 15 ev	L30	<b>Convention of chemical potential &amp; Dimerization in solution</b>	
Nov. 18	L31	<b>Virial coefficients 1 - derivations</b>	Dill, Ch15-16
Nov. 20		<b>Virial coefficients 2 - examples; Liquid State 1 - Basic Intro of <math>g(r)</math></b>	
Nov. 22	L32	<b>Mid-term Exam 2 (Chemical equilibrium; Lattice Vibration; Phase tra</b>	Chandler, Ch6
Nov. 25	L33	<b>Review of the second exam</b>	Chandler, Ch6
Nov. 27	L34	<b>Liquid State 2 - Thermodynamics and <math>g(r)</math></b>	McQuarrie, Ch12
Nov. 28-Nov. 30		<b>Thanksgiving</b>	
Dec. 2	L35	<b>Liquid State 3 - Chemical Potential; thermodynamic integration</b>	Chandler, Ch7; McQuarrie, Ch13
Dec. 4	L36	<b>Liquid State 4 - Perturbation theory; vdW theory</b>	Chandler, Ch7; McQuarrie, Ch13
Dec. 6	L37	<b>Liquid State 5 - Debye-Huckel 1</b>	Hill/McQuarrie
Dec. 9	L38	<b>Liquid State 6 - Debye-Huckel 2</b>	Hill/McQuarrie
Dec. 11	L39	<b>Polymers - a few relations</b>	Dill
Dec. 13	L40	<b>Final review</b>	
Dec. 20		<b>Final Exam (Cumulative)</b>	
<b>Other possible top</b>		<b>Monte Carlo I: Metropolis scheme</b>	
		<b>Monte Carlo II (umbrella sampling); imperfect gas</b>	